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## Model for calcite spherulite formation in organic, clay-rich, lacustrine carbonate shales (Barbalha Formation, Aptian, Araripe Basin, NE Brazil)

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1 Marine and Petroleum geology

2  
3 Model for calcite spherulite formation in organic, clay-rich, lacustrine  
4 carbonate shales (Barbalha Formation, Aptian, Araripe Basin, NE Brazil)

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23  
24 **Keywords**

25 Spherulites – Organic matter – Pre-Salt Reservoir Analogue – Microbialites – Barbalha

26        **Abstract**

27        The formation of carbonate spherulites, recognized as part of the South-Atlantic Pre-Salt  
28 reservoir rocks, remains enigmatic. Well-chosen analogues provide insights into depositional  
29 and diagenetic conditions and the biotic versus abiotic processes that might influence or even  
30 control spherulite growth. Radial fibrous calcite spherulites of the Batateira beds in the  
31 Barbalha Formation (Aptian, Araripe, NE Brazil) formed mainly in organic and clay-rich  
32 laminite layers as nodules that grew around different nuclei, now represented by, amongst  
33 others, micrite and apatite ostracods. They vertically alternate with clays, but also carbonate  
34 laminites, which locally also contain spherulites. Characterization of the organic matter content  
35 of spherulites (petrography, total organic carbon, vitrinite reflectance, stable isotopes) reveals  
36 TOC's of up to 30% related to the presence of mainly oil-prone alginite next to huminite,  
37 inertinite, sporinites and zooclasts. The complex mixture of clays, dominantly autogenic mixed  
38 layered illite-montmorillonite, which alternate with lacustrine carbonates and evaporites, is  
39 diagnostic for lakes in a semi-arid environment. The occurrence of gypsum, alveolar  
40 honeycomb spherulite textures, preferential concentration of pyrite at the spherulite-matrix  
41 contact and inward spherulite perforations suggest bacterial activity under shallow and  
42 evaporitic syn-depositional conditions. Although the Barbalha spherulites formed chemically-  
43 driven in a viscous matrix, the microbial influence, mainly as lamalginite in the clayey matrix,  
44 is undeniable. The mixed organic-silicate gel catalyzed calcite spherulite formation. The  
45 Barbalha spherulites consequently illustrate the complementary influence of clays, organic  
46 matter and evaporitic syn-sedimentary conditions.

47

48        **1. Introduction**

49        Lacustrine carbonate reservoirs from the Cretaceous South Atlantic basins, including  
50 spherulites, are considered non-conventional and poorly understood, among others, with

51 regard to biotic versus abiotic precipitation (e.g. Muniz and Bosence, 2015). Despite being one  
52 of the most prolific reservoirs discovered in recent times, the number of publications on the  
53 characterization of these Pre-Salt reservoir rocks, until 2017, remained strongly limited (Dias,  
54 1998; 2005; Terra et al., 2010; Wright, 2012; Rezende and Pope, 2015; Schröder et al., 2015;  
55 Thompson et al., 2015; Saller et al., 2016). A major barrier on the effective studying of the  
56 reservoir rocks was the lack of suitable deposits that can be investigated beyond proprietary  
57 and confidential drill core material of the South Atlantic (cf. Rogerson et al., 2017). The recent  
58 public release of the first core material and associated data, resulted in a strong increase in  
59 publication output on the Cretaceous lacustrine South-Atlantic deposits (e.g. Ceraldi and  
60 Green, 2017; Herlinger et al., 2017; Althaus et al., 2019; Bertolini et al., 2019; Farias et al.,  
61 2019; Lima et al., 2019; Amarante et al., 2020; Pietzsch et al., 2020; Pocas Belila et al., 2020;  
62 Wright and Barnett, 2020).

63         Due to the ultra-deep scenario of exploration and the associated lack of continuous pre-  
64 salt outcrops, reservoir information on lateral and vertical variation between seismic and core-  
65 scale remains a problem and highlights the importance of reservoir analogues. While it is  
66 difficult, if not impossible, to find analogues that cover the entire depositional and diagenetic  
67 complexity of the Pre-Salt deposits, worldwide fieldwork, experimental and modelling studies  
68 are able to provide insights addressing specific points of view, i.e. on the microbial or abiotic  
69 depositional processes (e.g. Bahniuk et al., 2015; Catto et al., 2016; Bouton et al., 2016; Chafetz  
70 et al., 2018; Kirkham and Tucker, 2018), the sedimentology (e.g. Seard et al., 2013; Awramik  
71 and Buchheim, 2015; Ronchi and Francesco, 2015; Rogerson et al., 2017; Mercedes et al.,  
72 2017; Valente et al., 2017), diagenesis (e.g. Gindre-Chanu et al., 2016; Claes et al., 2017a; De  
73 Boever et al., 2017), geochemistry (e.g. Teboul et al., 2016; 2017; 2018; Tutolo and Tosca,  
74 2018; Claes et al., 2019), petrophysical characteristics (e.g. Muniz and Bosence, 2015; Soete  
75 et al., 2015, 2017; Claes et al., 2017b; Regnet et al., 2018; Hosa et al., 2019) or structural

76 aspects (Van Noten et al., 2013, 2019; Miranda et al., 2018; Török et al., 2019). Only the  
77 assembly and integration of information from different analogue sites with the core, sidewall  
78 core, borehole, well and seismic reservoir data leads to a proper understanding of the Pre-Salt  
79 reservoirs to support their exploitation (Virgone et al., 2013). The success of the assembly  
80 strategy for understanding the Pre-Salt and its economic value depends on selecting and  
81 studying suitable analogues.

82         The Aptian lacustrine deposits in the Barbalha formation of the Araripe Basin are  
83 almost time-equivalent to the Pre-Salt reservoirs and contain spherulites, the major focus of  
84 this study. To avoid confusion, distinction in terminology is made between the individual  
85 (sub)spherical allochems with smooth or lobated rims, i.e. the spherulites, and the rock  
86 containing spherulites, i.e. a spherulite (Terra et al., 2009). The Araripe spherulites are best  
87 developed in the Batateira Beds and are described as “*peloids in paperthin shales*” (Chagas et  
88 al., 2007) and “*millimetric to centimetric microbial carbonate laminae and nodules*” (e.g.  
89 Scherer et al., 2015; Fambrini et al., 2019). The Batateira beds, named by Hashimoto et al.  
90 (1987), are defined as plumbiferous (Farin, 1974), black bituminous shales with carbonate  
91 sheets of algal origin, coprolites, ostracods and fish remains (Assine, 1992 and 2007; Assine et  
92 al., 2014) and reflect deposition in an anoxic, low energy lacustrine system (Scherer et al.,  
93 2015; Fambrini et al., 2019) in a semi-arid region (Lima and Perinotto, 1984). The Batateira  
94 beds occur as an important stratigraphic marker of 10m thick black shale-carbonate mixed  
95 interval continuous over at least 20km in the Araripe Basin (Chagas et al., 2007; Fabrini et al.,  
96 2019), but are possibly even continuous into neighboring basins (see discussion including  
97 Varejao et al., 2016). Spigolon et al. (2015) evaluated the petroleum composition and  
98 confirmed the source rock potential of the Batateira spherulites.

99         This study aims to provide a sedimentological characterization of the Barabalha  
100 lacustrine spherulites in order to understand their depositional and early diagenetic syngenetic

101 conditions and formation mechanisms. The basis is a multi-disciplinary approach, including  
102 fieldmapping and logging, petrography, TOC, vitrinite reflectance and carbonate and organic  
103 matter stable isotope analyses. The authors, besides outcrops of the Barbalha Formation, visited  
104 lacustrine outcrops of the covering formations, mainly the Crato Formation, in order to be able  
105 to consider the observations in a larger frame. Given the ongoing discussions on the microbially  
106 mediated versus the dominantly abiotic silica gel nucleated origin of the Barra Velha and  
107 Macabu Pre-Salt spherulites (e.g. Dorobek et al., 2012; Wright and Barnett, 2015; Kirkham  
108 and Tucker, 2018; Wright and Barnett, 2020) and the discussion of an open deep lake versus  
109 closed shallow lake scenario (Mercedes-Martin et al. 2019 and Wright and Barnett 2020,  
110 respectively), the Barbalha Formation might form a geographically proximal and possibly  
111 genetic analogue, and/or can provide additional information on the settings in which spherulites  
112 can form. This is not only of use for the Pre-Salt hydrocarbon reservoirs but also for other  
113 worldwide spherulite deposits, like, for example, in the Precambrian of Arctic Norway  
114 (Meinhold et al., 2019).

115

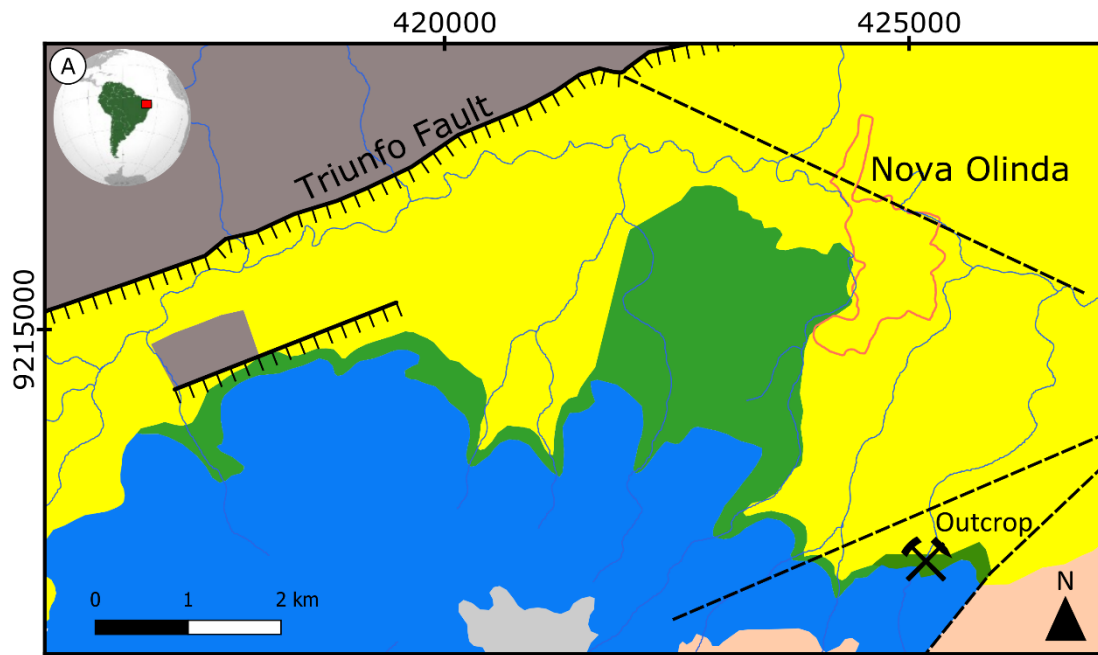
## 116 **2. Geological setting**

117 The study area is located at the northern border of the Araripe Basin (Fig. 1), which is  
118 the largest intraplate basin of Northeast Brazil (Castro and Castelo Branco, 1999). The tectonic  
119 evolution of this basin is related to the break-up of Gondwana and subsequent South Atlantic  
120 rifting (Matos, 1999). The underlying basement rocks are dominantly Archean orthogneisses  
121 (Granjeiro Complex), ranging in composition from quartz diorite to granite, but meta-  
122 ultramafics, meta-gabbros, amphibolites and Neoproterozoic meta-sedimentary rocks  
123 (Cachoeirinha Group) locally occur, as well as a Cambrian granitic intrusion (Celestino et al.,  
124 2020). The basin has an E-W-elongated shape and is bounded to the north by the Triunfo Fault,  
125 which represents the brittle reactivation of the Patos shear zone (Celestino et al., 2020).

126           The stratigraphy of the Araripe Basin can be divided into five tectono-sedimentary  
127 sequences, namely: a) Paleozoic, represented by the Ordovician-Devonian continental deposits  
128 of the Cariri Formation; b) prerift, represented by the Tithonian continental deposits of the  
129 Brejo Santo and Missão Velha Formations; c) rift, composed of the Neocomian continental  
130 deposits of the Abaiara Formation; d) postrift I, known as the Santana Group which includes  
131 the Aptian (Arai and Assine, 2020) continental deposits of the Barbalha (before referred to as  
132 the Rio da Batateira; Ponte and Appi, 1990), Crato, and Ipubi formations and the lacustrine to  
133 marine deposits of the Romualdo Formation; and e) postrift II, which includes the Albian-  
134 Cenomanian continental deposits of the Araripina and Exu formations (Assine, 2007; Assine  
135 et al., 2014) (Fig.1).

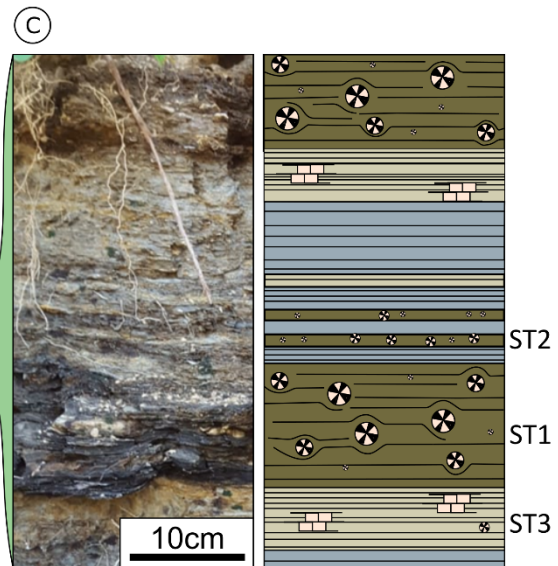
136           This study focusses on the lacustrine spherulites of the early Aptian in the Barbalha  
137 Formation (P-270 biozone; Lima and Perinotto, 1984). The Barbalha Formation consists of  
138 two main sequences of fluvial sandstones to mudstones that grade into lacustrine black  
139 shales. The first sequence culminating in the black-shale-carbonate mixed interval of the  
140 Batateira beds, is unconformably interrupted by a renewed fluvial system (Hashimoto et al.,  
141 1987; Assine, 1992; Chagas et al., 2007; Assine, 2007; Scherer et al., 2015; Fambrini et al.,  
142 2019). The upper sequence of the Barbalha Formation grades transitionally into the Crato  
143 Formation (Fambrini et al., 2019) which represents the main lacustrine phase of the continental  
144 succession of the Araripe Basin (Neumann et al., 2003). Fieldwork included a new outcrop of  
145 the Barbalha Formation in the northern part of the Araripe Basin (Fig. 1), in addition to the  
146 stratigraphic logs previously studied by a.o. Scherer et al. (2015) and Fambrini et al. (2019).  
147 Sampling for further analyses focused on the Batateira beds of the Barbalha Formation from  
148 the outcrop indicated in Figure 1.

149



(B)

Chronostratigraphic units			Tectonic sequence	Lithostratigraphic units	
International	Local			Formation	Group
Upper Cretaceous	Cenomanian		Post-rift II		Araripe
				Araripina	
Lower Cretaceous	Albian	Alagoas	Post-rift I	Romualdo	Santana
	Aptian			Ipubi Crato	
				Barbalha	
Paleozoic	Silurian-Devonian		Pre-rift	Cariri	
Pre-cambrian	Neo-archean		Basement		Granjeiro complex



- (D)
- Colluvial deposits
  - Normal Fault
  - Inferred Faults
  - Drainages
  - City
  - Laminite
  - Recrystallisation
  - Organic-rich shale
  - Spheroids
  - Marly transition between laminite and shale

150

151

152

153

*Figure 1: A) Geological map of the area around the sampled outcrop in the north of the Araripe Basin. B) Stratigraphy of the encountered lithologies in A. C) Representative*



154 *outcrop section of the Batateira Beds in the middle of the Barbalha Formation with sample*  
155 *types indicated; ST1:Organic-rich carbonate shale spherulitites with micritic nuclei (Fig. 2).*  
156 *ST2:Organic-rich carbonate shale ostracod-rich spherulitites (Fig. 3). ST3: Carbonate*  
157 *laminite with spherulites (Fig. 5). Legends of A and C.*

158

### 159 **3. Methods**

#### 160 *3.1 Petrography*

161 Because of the fragility of some of the samples, all were doubly impregnated under  
162 vacuum with resin before thin-section (KU Leuven) and polished section (RWTH Aachen)  
163 preparation. A fluorescent dye was used in order to easily distinguish (micro-)porosity with  
164 incident fluorescent light microscopy. Polishing was done with oil to avoid clay swelling. Thin-  
165 sections were scanned with parallel and crossed polarized transmitted light and incident light  
166 at a resolution of 600 dpi on an EPSON scanner with an EPSON film adapter (Seiko Epson  
167 Corporation, Nagano, Japan) at the KU Leuven. Petrographic characterization (KU Leuven)  
168 was conducted on an Olympus BX60 (Olympus Corporation, Tokyo, Japan) and Leica DM LP  
169 Parallel (Leica Microsystems, Wetzlar, Germany) and Crossed Polarizer Optical and  
170 Fluorescence light (PPL, CPL and FL respectively) microscope. Cold cathodoluminescence  
171 (CCL) observations were carried out using a Zeiss MK5-2 microscope under 10 KeV and 250  
172 mA. Reflectance measurements (RWTH Aachen) were performed using a Zeiss Axioplan  
173 incident light microscope at  $\lambda = 546$  nm equipped with a 50 $\times$ /0.85 Epiplan-NEOFLUAR oil  
174 immersion objective. Random huminite reflectance ( $HR_r$ ) measurements were conducted  
175 according to the standard procedure described in Taylor et al. (1998). Each measurement was  
176 calibrated with a Leuco Sapphire standard ( $VR_r=0.592\%$ ). Samples for SEM (RWTH Aachen)  
177 were Ar-ion polished using a Leica EM TIC3X Broad Ion Beam (BIB) polisher with a rotary  
178 stage at low angle with respect to the sample surface and 6 kV acceleration voltage for 1h,

179 followed by a tungsten coating of ~7.5nm to make the specimen conductive for high-resolution  
180 imaging. Zeiss Supra 55 FE and FEI Versa 3D Dual Beam field emission scanning electron  
181 microscopes (SEM) were used for imaging. Both microscopes were equipped with secondary  
182 electron (SE), backscattered electron (BSE) and energy dispersive X-ray (EDX) detectors,  
183 allowing for the simultaneous collection of image data and element concentrations to identify  
184 different minerals.

185

### 186 3.2 *QEMSCAN*

187 QEMSCAN (Quantitative Evaluation of Materials by SCANning electron microscopy),  
188 (FEI/Thermo corporate headquarters, North America Nanoport, Hillsboro, OR, USA) analysis  
189 was applied with a beam voltage of 15.0 keV and a Bruker XFlash 5030 detector (RWTH  
190 Aachen). The horizontal and vertical point spacing (height and width of a pixel) is 4.76  $\mu\text{m}$ .  
191 The requested field size is 1500  $\mu\text{m}$  (width and height each). The total field size is 29 mm  
192 height and 20 mm width. 2000 X-Ray counts were detected per pixel.

193

### 194 3.3 *X-Ray Diffraction (XRD)*

195 The full XRD analyses were performed at RWTH Aachen. Rock samples were crushed  
196 manually in a mortar and subsequently milled with a McCrone Micronising mill for 20 minutes  
197 to ensure uniform crystallite sizes. Milling was done in ethanol to avoid dissolution of water-  
198 soluble components and strain damage to the samples. An internal standard (Corundum, 20 wt.  
199 %) was added to improve the accuracy of the analysis. Bulk mineralogical compositions were  
200 derived from X-ray diffraction patterns measured on randomly oriented powders. For textured  
201 clay samples, the clay fraction was separated with the Atterberg method after removal of  
202 carbonates, organic matter and iron oxides (procedures from Moore and Reynolds, 1997 and  
203 Lagaly et al., 1997). ~100 mg clay sample was dispersed in 3 mL H<sub>2</sub>O. Two suspensions are

204 made from the Mg-exchanged clay fraction. To one of them two to three drops of glycerol or  
205 a drop of glycerin were added. Both suspensions were ultrasonically treated for >10 minutes  
206 in order to obtain a stable suspension. The suspensions are spread onto porous ceramic, glass  
207 or silica mono-crystal plates. For both bulk and oriented samples, the measurements were done  
208 on a Bruker AXS D8 Advance diffractometer using CuK $\alpha$ -radiation produced at 40 kV and 40  
209 mA and a Huber Guinier diffractometer using CoK $\alpha$ -radiation at 40 kV and 30 mA. The  
210 divergence slit is fixed at 1 mm, with a secondary Soller slit behind the sample. The detector  
211 is a scintillation counter with graphite monochromator. Diffractograms of bulk samples were  
212 recorded from 2° to 92° 2 $\theta$  in 0.02° steps and a counting time of 5 s. Textured samples are  
213 complemented by measurements of glycerol added and 500 °C heated samples with  
214 diffractograms recorded from 2° to 40° 2 $\theta$  in 0.02° steps and a counting time of 5 s.

215 Quantitative phase analysis was performed by Rietveld refinement using Profex 4.0.0  
216 BGMN software, with customized clay mineral structure models (Ufer et al., 2008).

217

### 218 3.4 Total Organic Carbon (TOC)

219 15 mg of each sample were weighed into 9 × 5 mm Ag cups, and carbonates were  
220 removed by repeated acidification with diluted (2%) HCl. For the determination of %OC  
221 (organic carbon), %N (nitrogen) and the stable carbon isotope composition of the OC fraction  
222 ( $\delta^{13}\text{C}_{\text{OC}}$ ), samples were combusted in an elemental analyser–isotope ratio mass spectrometer  
223 (EA-IRMS, ThermoFinnigan Flash HT, and ThermoFinnigan DeltaV Advantage), and data  
224 were calibrated using an in-house Leucine and IAEA-C600 standard (caffeine) at the KU  
225 Leuven. Two samples were selected for exploratory Rock-Eval analysis (RWTH Aachen),  
226 using a Girdel Rock-Eval MK II pyroanalyzer.

227

### 228 3.5 Stable isotopes

229 Samples for stable oxygen and carbon isotopes were analysed at the KU Leuven on a  
230 Thermo Delta V Advantage isotope ratio mass spectrometer coupled to a GasBench II.  
231 Spherulites were handpicked from their matrix before being crushed. Samples were flushed  
232 with helium and reacted with 100% phosphoric acid to produce CO<sub>2</sub> gas and were allowed to  
233 react for 24 h at 25 °C to reach isotopic equilibrium. Data from each run were corrected using  
234 the regression method with LSVEC ( $\delta^{18}\text{O} = -26.7\text{‰}$ ,  $\delta^{13}\text{C} = -46.6\text{‰}$ ), NBS-19 ( $\delta^{18}\text{O} =$   
235  $-5.01\text{‰}$ ,  $\delta^{13}\text{C} = -23.2\text{‰}$ ), and NBS-19 ( $\delta^{18}\text{O} = -2.2\text{‰}$ ,  $\delta^{13}\text{C} = +1.95\text{‰}$ ) as standards, as well  
236 as using two in-house CaCO<sub>3</sub> standards, which were regularly calibrated against NBS-19 and  
237 LSVEC. Long-term standard deviations were better than 0.1‰. Both  $\delta^{18}\text{O}$  and  $\delta^{13}\text{C}$  values of  
238 samples are expressed relative to Vienna-Pee Dee Belemnite (V-PDB).

239

### 240 3.6 Computer Tomography (XCT)

241 XCT image acquisition was performed at the KU Leuven on a Phoenix Nanotom S (GE  
242 Measurement and Control Solutions, Wunstorf, Germany), equipped with a 180 kV/15 W  
243 nanofocus X-ray tube and a 2304 x 2304 pixels Hamamatsu detector, with a 5  $\mu\text{m}$  pixel pitch.  
244 An acceleration voltage of 70 kV and a current of 500  $\mu\text{A}$ , in nanofocus mode 0, were applied  
245 to generate the X-rays using a tungsten target. A 0.1 mm thick copper filter was installed during  
246 the scans. Radiographic projections were acquired for a 360° rotation sector, with a 0.2°  
247 rotation step. The exposure timing was set to 500 ms. The acquisition time was only 15 minutes  
248 in fast-scan mode, which allows for a continuous sample rotation. The filtered-back projection  
249 algorithm in the Phoenix dat|x 2 reconstruction software (GE Measurement and Control  
250 Solutions, Wunstorf, Germany) was used to reconstruct all the sectional planes of the generated  
251 image stack, until finally a 3D model of the sample was generated. The resulting 3D model has  
252 an isotropic voxel size of 9  $\mu\text{m}$ . The XCT images were segmented, visualized and analysed in  
253 Avizo 2019.1 (Thermo Fisher Scientific, Hillsboro, Oregon, USA).

254

## 255 **4. Results**

### 256 *4.1 Spherulites in organic-rich carbonate shales*

257         The Araripe spherulites (mm up to cm scale) most commonly occur in the clayey (clay  
258 content reaches up to 50%), exceptionally organic rich (TOC up to 30%), anastomosingly  
259 laminated shaly carbonate matrices (Figs. 2 and 3) of the Batateira beds. The spherulites consist  
260 of radial fibrous calcite crystals with intercrystalline microporosity (e.g. Fig. 3H-K). Their  
261 nuclei can be composed of anhedral micrite (Fig. 2) but are also related to other substrates, like  
262 apatite-calcite ostracods in ostracod-rich lithologies (Fig. 3). Both valves of the ostracods occur  
263 together. Under crossed polar configuration, the spherulites show the typical cross extinction  
264 (Fig. 2D & F and 3B, D & G). The spherulites and ostracods are bright yellow to light brown  
265 under Cold Cathodoluminescence as opposed to their dark matrix (Fig. 2G, 3E & H). The  
266 spherulite size, even within a hand sample is highly variable (Fig. 2 and 3, in particular 2N)  
267 and observations from classical petrography and SEM show growth competition, as illustrated  
268 in Figure 2H. The matrix laminae bend and wrap around the spherulites.  $\mu$ CT revealed  
269 perforations from the edge towards the inside of the spherulites (Fig. 2K). At the contact  
270 between the spherulites and the matrix, siderite, iron oxides (goethite) and sometimes pyrite  
271 (Fig. 2L) are concentrated. Pyrites are also observed to less extent inside the matrix of the  
272 spherulites. The clay fraction of the spherulite samples is a complex mixture of dominantly  
273 illite, smectite (mainly montmorillonite), chlorite and kaolinite (Fig. 4, Table 2). The main  
274 phase has been determined as interstratified illite – montmorillonite (Fig. 4). Gypsum and  
275 minor amounts of siderite occur in association with the laminated shale matrix together with  
276 clay to silt-sized detrital quartz and feldspar. Furthermore, these spherulite-rich beds are  
277 associated with larger centimeter to decimeter scale carbonate concretions. The matrix-based  
278 organic matter mainly consist of amorphous organic matter (AOM) from algae (alginite),

279 which corresponds to highly oil-prone type I kerogen (Hydrogen Index > 500 mg HC/ g TOC,  
280  $T_{\max} > 400^{\circ}\text{C}$ ). Within the fabrics of the spherulites, alveolar honeycomb features are observed  
281 (Fig. 2J).

282 The Barbalha shaly matrix surrounding the spherulites display  $\delta^{18}\text{O}$  signatures that vary  
283 from -8 to -7‰ and from -3 to -5‰ for  $\delta^{13}\text{C}$  and  $\delta^{13}\text{C}_{\text{org}}$  of -24 to -29‰ (Table 1). The  
284 relatively low amount of associated nitrogen (0.5-0.8%) results in high C/N ratios of 33 to >36.  
285 Successful  $\delta^{15}\text{N}_{\text{org}}$  measurements varied between 0 and 4‰ (Table 1).

286 The spherulites show similar  $\delta^{18}\text{O}$  signatures to their matrix (-8 to -6‰). In most cases  
287 the  $\delta^{13}\text{C}$  signatures are slightly lower (-4 to -5‰), with exception of a few spherulites in an  
288 ostracod matrix (<-6‰; ~2‰ difference; Table 1). The  $\delta^{13}\text{C}_{\text{org}}$  varies between -25 and -22‰.  
289 The nitrogen reaches maximally 0.13%, resulting in C/N ratios from 6 to >37.  $\delta^{15}\text{N}_{\text{org}}$  varies  
290 between 0 and 2.8‰ (Table 1).

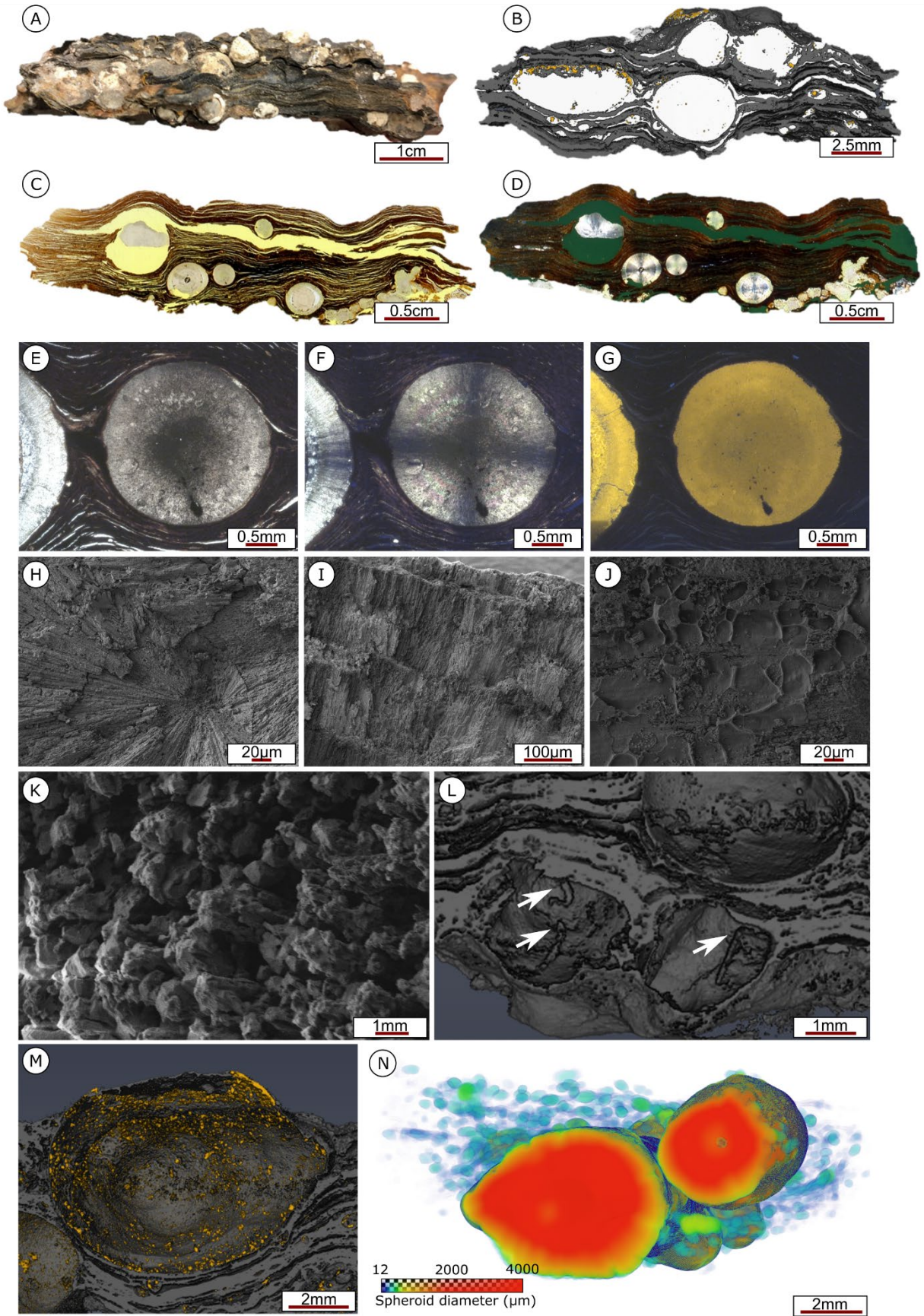
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#### 292 4.2 *Spherulites in carbonate laminites*

293 The Barbalha laminites are low porous carbonate mudstones that show a macro-scale  
294 lamination of beige and greyish-brownish carbonate laminae (0.3-3mm thick) and locally  
295 contain crustiform to spherical concretions (Fig. 5). Massive horizontally elongated  
296 concretions with concentric growth laminae can be distinguished from agglomerative and  
297 individual radial fibrous spherulites. Laminae anastomose around the concretions and  
298 spherulites. On micro-scale the matrix consists of peloidal micrite ghost grains that  
299 recrystallized into microspar. The laminites occur as laterally continuous, few to tens of  
300 centimeter thick, intervals intercalated with the shaly spherulites. With a TOC of ~2%, the  
301 organic matter content of the laminites is considerably lower compared to the shaly  
302 spherulites. The laminites and the spherulites inside consist of almost purely (>90%) calcite  
303 (Table 2). Darker laminae are associated with iron and/or manganese oxides (Fig. 5). The latter

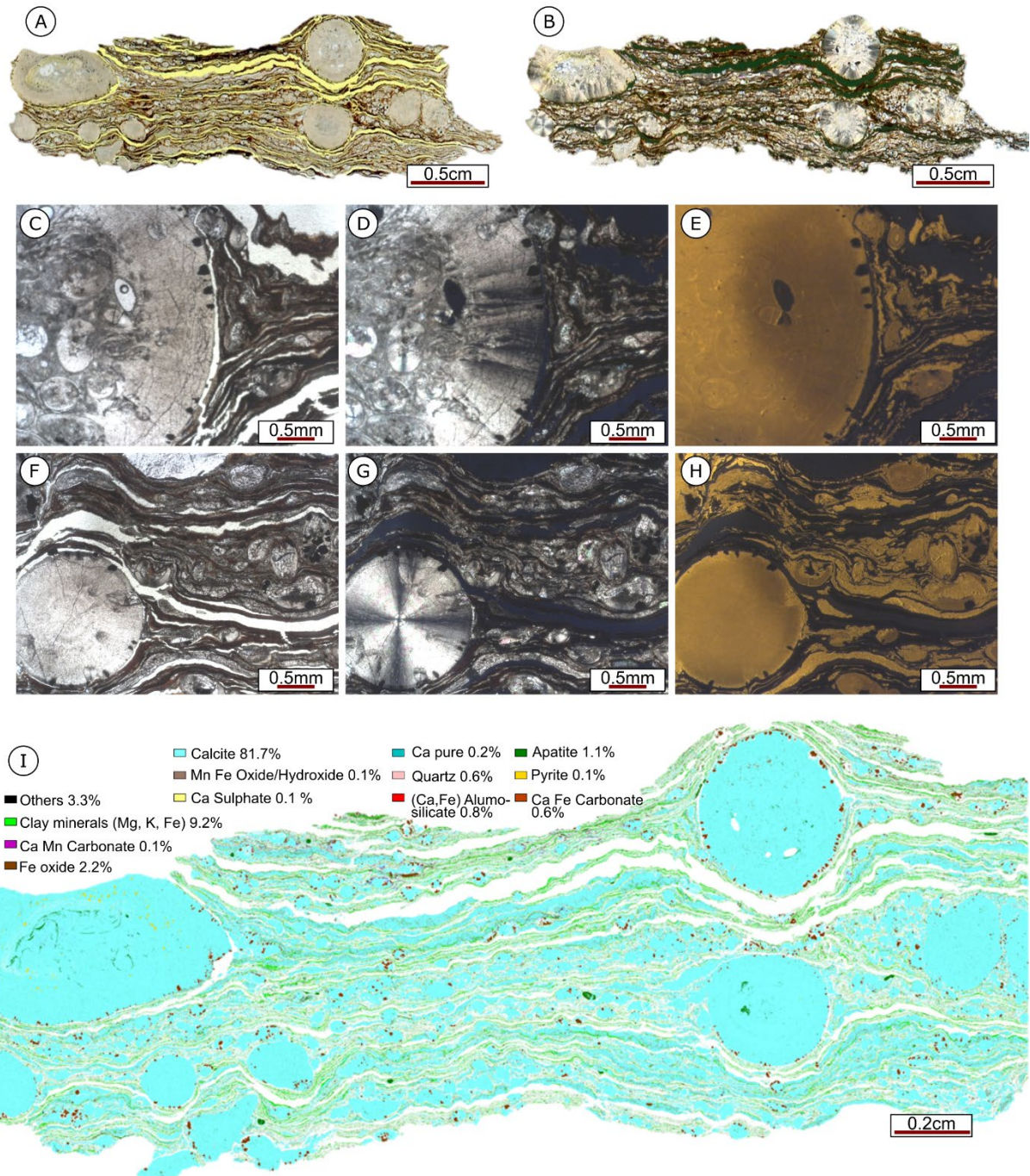
304 also occur preferentially at the spherulite-matrix interface and, more inward the spherulite  
305 concretions, some pyrite is observed. Apatite occurs related to local organic matter lenses and  
306 fossil fragments. Minor amounts of clay, quartz and feldspar grains are identified from XRD  
307 and QEMSCAN.

308         The spherulite  $\delta^{18}\text{O}$ - $\delta^{13}\text{C}$  values (-7.8‰ for both) differ from those of their laminite  
309 matrix ( $\delta^{18}\text{O}$  of -7‰ and -2‰ for  $\delta^{13}\text{C}$ ). The laminite matrices, with 0.08-0.09% nitrogen,  
310 have a C/N ratio of 19.7 to 23.1 and  $\delta^{13}\text{C}_{\text{org}}$  -25‰ and  $\delta^{15}\text{N}_{\text{org}}$  of 4.5 to 5 (Table 1).





312            *Figure 2: Images of spherulites with micritic cores in an organic-rich (up to 30%),*  
313 *clay-rich laminites. A) Outcrop spherulitite sample with wavy lamination of the dark matrix*  
314 *surrounding white spherulites. B) Rendered  $\mu$ CT image with segmentation of the different*  
315 *phases, i.e. calcite in white, matrix in black and iron oxides and/or pyrite in yellow. C) and D)*  
316 *Thin-section scans of spherulitite sample with micritic cores under PPL and CPL, respectively.*  
317 *Notice the typical cross extinction of the spherulites under CPL. E), F) and G) Micro-*  
318 *petrographic images of spherulites in the organic-rich, clay-rich matrix under PPL, CPL and*  
319 *CCL, respectively. Notice how the laminae of the matrix bend around and wrap the spherulites.*  
320 *H) SEM image illustrating the competitive growth of radial fibrous spherulites. I) SEM image*  
321 *showing the different growth layers of fibrous crystals forming the spherulite. J) SEM image*  
322 *showing hexagonal / alveolar honeycomb texture within the spherulite. K) SEM image*  
323 *illustrating the intercrystalline microporosity related to the spherulites. L) Rendered  $\mu$ CT*  
324 *image of the segmented matrix. Removal of the segmented spherulites reveals perforations*  
325 *from the edge inwards the spherulites. M) Rendered  $\mu$ CT image of the segmented matrix and*  
326 *iron oxides and/or pyrite. Removal of the segmented spherulites emphasizes the preferential*  
327 *occurrence of the iron oxides/sulfides on the contact between the spherulites and their matrix.*  
328 *N) Rendered  $\mu$ CT image of zone within a spherulitite sample with spherulites colored*  
329 *according to size.*



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Figure 3: Spherulites with ostracod nucleus. A) and B) Thin-section scans of spherulite sample rich in ostracods under PPL and CPL, respectively. C), D) and E) Micro-petrographic images of spherulite edge with ostracods in its nucleus and surrounding matrix in PPL, CPL and CCL, respectively. Notice how the laminae bend around the spherulite. F), G) and H) Micro-petrographic images of spherulite in an ostracod rich matrix in PPL, CPL and CCL, respectively. I) Results of QEMSCAN analysis of the thin-section shown in A and B

337 with mass percentages of minerals given. Note that despite the use of oil for thin-section  
 338 preparation, part of the matrix (mainly clays and organic matter) was lost.

339

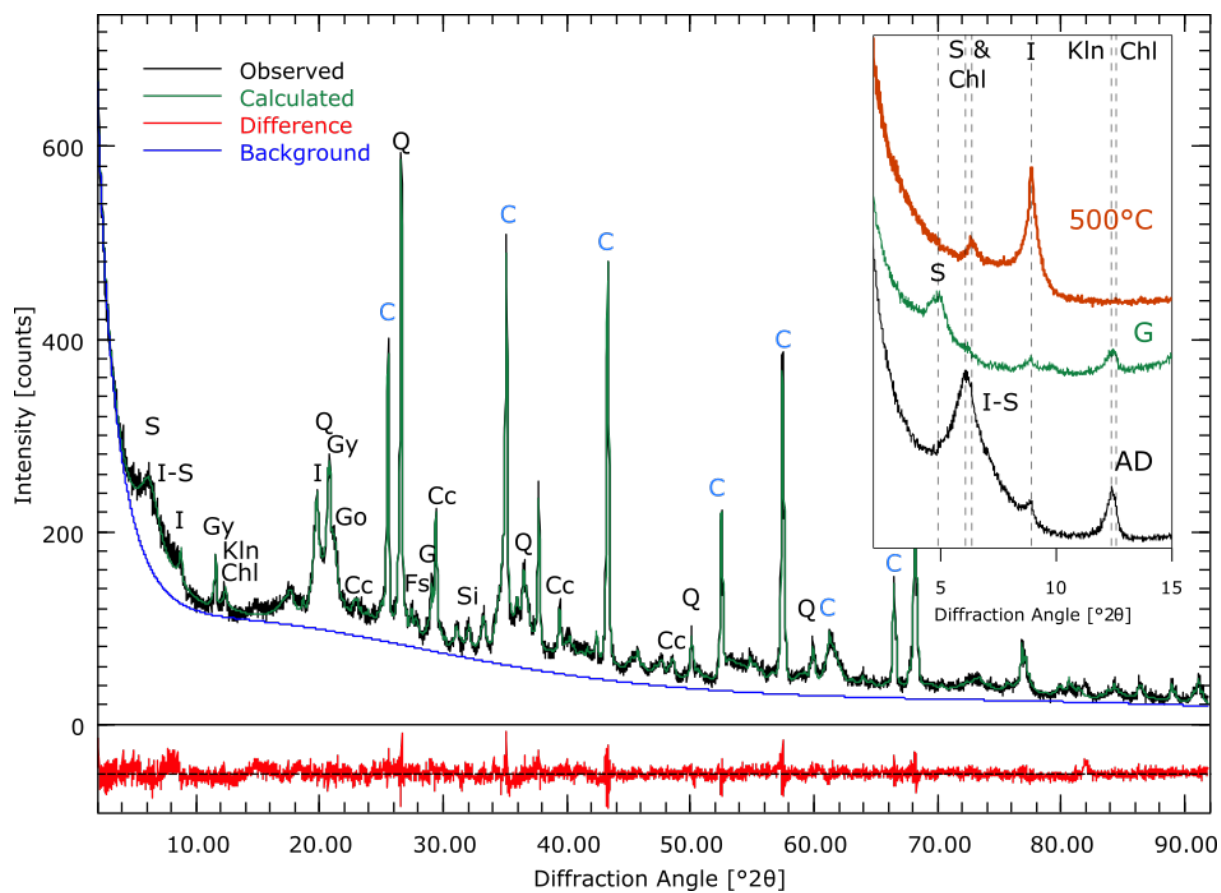
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341 *Table 1: Results geochemical analyses (ND = not determined).*

Sample	Lithotype / Description	%C	%N	C/N	$\delta^{13}\text{C}_{\text{Org}}\text{‰}$	$\delta^{15}\text{N}_{\text{Org}}\text{‰}$	$\delta^{13}\text{C}\text{‰}$	$\delta^{18}\text{O}\text{‰}$
Araripe1	Shaly spherulitite matrix	27.17	0.79	34.57	-24.7	4.0	-3.33	-7.96
Araripe2	Shaly spherulitite matrix	28.99	0.80	36.08	-24.3	3.7	ND	ND
Araripe3	Spherulite in shaly matrix	0.37	0.06	6.67	-24.1	2.4	-4.05	-7.67
Araripe4	Spherulite in shaly matrix	5.01	0.13	37.48	-22.1	ND	-4.74	-7.61
Araripe5	Spherulite in shaly matrix	ND	ND	ND	ND	ND	-3.67	-6.99
Araripe6	Spherulite in shaly matrix	ND	ND	ND	ND	ND	-4.63	-8.22
Araripe7	Spherulite in shaly matrix	ND	ND	ND	ND	ND	-4.91	-7.63
Araripe8	Shaly spherulitite matrix with ostracods	22.28	0.66	33.61	-28.8	ND	-4.97	-6.95
Araripe9	Shaly spherulitite matrix with ostracods	22.93	0.67	34.46	-28.5	0.0	-5.09	-6.87
Araripe10	Shaly spherulitite matrix with ostracods	16.74	0.50	33.72	-28.7	0.0	-5.12	-6.62
Araripe11	Shaly spherulitite matrix with ostracods	18.38	0.53	34.84	-28.7	ND	-5.19	-6.82
Araripe12	Spherulite in ostracod-rich matrix	1.20	0.08	15.16	-24.8	0.4	-5.93	-6.69
Araripe13	Spherulite in ostracod-rich matrix	0.56	0.05	11.35	-24.1	2.8	-6.42	-7.03
Araripe14	Spherulite in ostracod-rich matrix	0.72	0.09	8.29	-24.3	ND	-6.29	-7.02

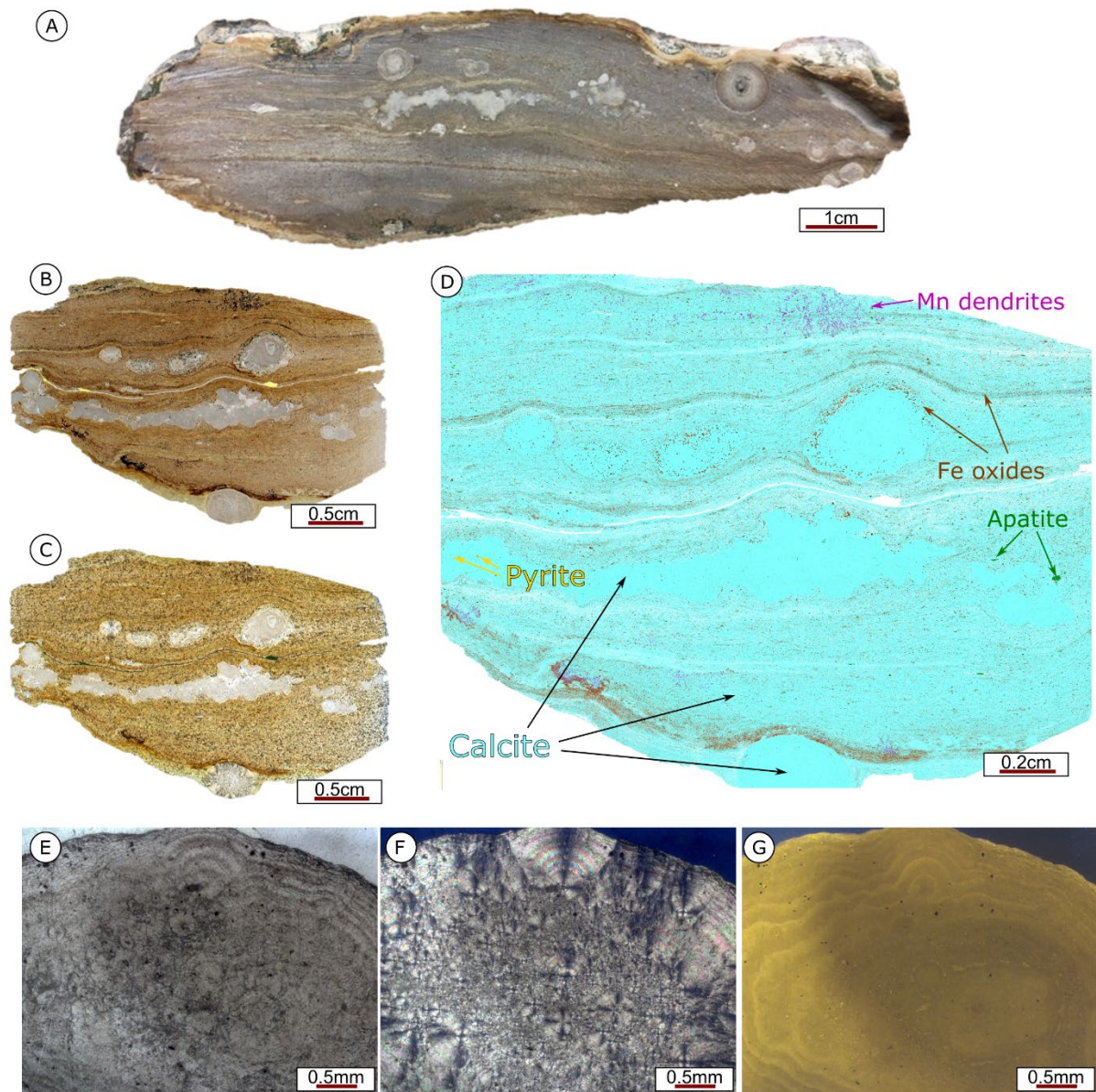
Araripe15	Spherulite in ostracod-rich matrix	0.37	0.06	6.04	-24.6	2.5	-4.28	-6.99
Araripe16	Spherulite in ostracod-rich matrix	ND	ND	ND	ND	ND	-5.51	-5.71
Araripe17	Spherulite in ostracod-rich matrix	ND	ND	ND	ND	ND	-4.81	-6.92
Araripe18	Laminite – spherulite matrix	1.77	0.09	19.74	-25.0	4.5	-2.22	-7.06
Araripe19	Laminite – spherulite matrix	1.95	0.08	23.10	-25.0	5.0	-2.08	-7.04
Araripe20	Grey-Blueish spherulite in laminite – spherulite	ND	ND	ND	ND	ND	-7.77	-7.83

342



343

344 *Figure 4: Representative XRD diffraction pattern for the shaly matrix of spherulites*  
345 *with micritic nucleus with inset of textured clay fraction diffraction patterns of Air Dried (AD),*  
346 *Glycerin added (G) and 500°C heated (500°C) samples. Together with the (060) reflection at*  
347 *1.5Å, the smectite was determined as mainly montmorillonite. Notice the high background due*  
348 *to the presence of organic matter in the sample. Quantification was further verified after*  
349 *removal of carbonates, organic matter and iron oxides. The main peaks of the minerals are*  
350 *highlighted for internal standard Corundum (C), Calcite (Cc), Chlorite (Chl), Feldspar (Fs),*  
351 *Goethite (Go), Gypsum (Gy), Illite (I), interstratified Illite-Smectite (I-S), Kaolinite (Kln),*  
352 *Quartz (Q), Smectite (S), Siderite (Si).*



353

354 *Figure 5: Spherulites in carbonate laminites. A) Hand specimen of laminite with*

355 *carbonate concretions and spherulites. B) and C) Thin-section scans of laminite sample with*

356 *concretions and spherulites under PPL and CPL, respectively. D) Interpreted mineralogical*

357 *QEMSCAN analysis of the thin-section shown in B and C. The main mineralogies are*

358 *indicated. The sample consists for >90% of calcite. Iron oxides occur along the laminae and*

359 *at the contact between spherulites and concretions. Pyrite occurs inward of the concretions.*

360 *Apatite is observed related to fossil fragments in the matrix. E), F) and H) Micro-petrographic*

361 *image of agglomerative spherulite in PPL, CPL and CCL, respectively.*

362

Table 2: Mineralogical results (wt%) of representative samples of the different

363 lithologies.

	Representative sample of:	Carbonate laminite – spherulitites matrix	Shaly spherulitite matrix	Shaly spherulitite matrix with ostracods	Spherulites
Mineralogy (%)	Quartz	0.1	7.3	1.4	0.4
	K-feldspar		1.5		
	Calcite	91	3.2	76.2	98.6
	Dolomite/Ankerite	0.1	0.1		
	Siderite	4.8	0.2	0.2	0.3
	Goethite	1.4	5.4	1.7	0.7
	Apatite	0.5		<1%	
	Gypsum		2.4	4.0	
	Total clay	0.8	51.0	2.9	0.0
	Kaolinite		2.2		
	Interstratified I-S		46.2	2.9	
	Chlorite		2.6		
	TOC (%)		~2	25-32	15-23

364

365 

### 4.3 Organic matter

366 The high TOC values stimulated an in-detail characterization of the organic matter in

367 the shaly spherulitite samples. Organic particles are composed of smaller, dispersed huminites

368 and isolated inertinites, occurring as inertodetrinite and fusinite (Fig. 6A and B). The huminite

369 particles show an average random reflectance of 0.32% (n = 72, st. dev. = 0.07), indicating a

370 low thermal maturity of the Araripe spherulite samples. Very abundant are fishbones and

371 calcious faunal remnants (Fig. 6C). The fishbones show an association with apatite (Fig. 6D).

372 There are some organic particles observed which can be interpreted as zooclasts. These share

373 similar petrographic features and are characterized by their roundish shape which shows a high

374 reflection similar to that of the inertinite (Fig. 6E and F). In most cases, the cavities enclosed

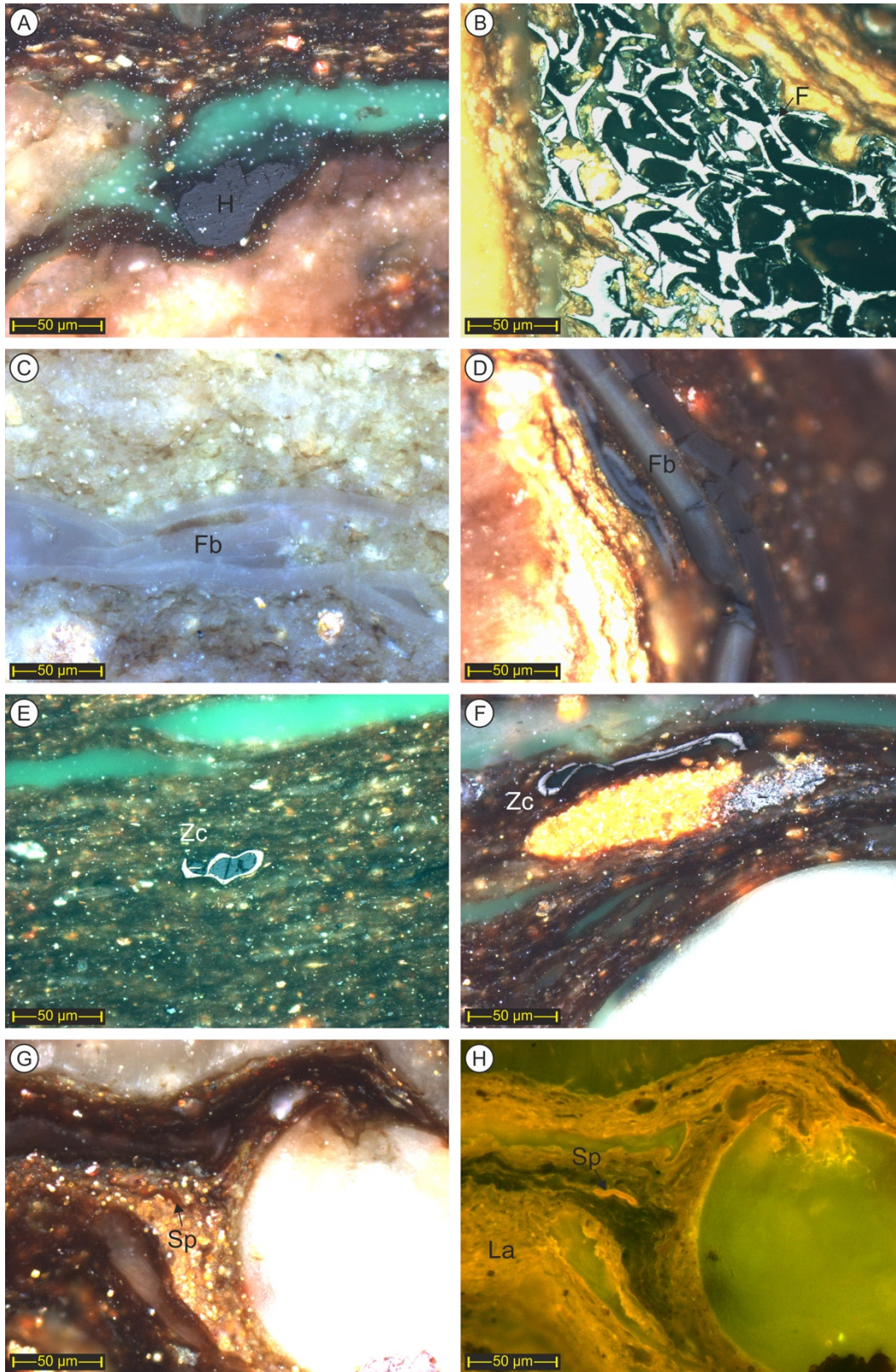
375 by them are filled with a substance petrographically similar to solid bitumen (Fig. 6E). This

376 substance shows a low reflectivity that similar to the huminites ( $VR_r = 0.29\%$ , n = 47, st. dev.

377 = 0.06). The exact origin of these particles, which are typical for the samples investigated,

378 requires further research. The observation of the samples in fluorescence mode allows for the  
379 better visualization of macerals of the liptinite group (Fig. 6G and H). The high contents of  
380 lamalginite, finely distributed in between the clay layers, are typical for lacustrine organic-rich  
381 sediments (Fig. 6H). Their dominant occurrence may be the main reason for the high TOC  
382 values of the samples. Sporinites (Fig. 6H) are occasionally found, as huminite and inertinite  
383 representing an aeolian input of higher land plant material.





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*Figure 6: Microphotographs focusing on the organic matter of the spherulites taken in incident white (A-G) and blue light (H). H = huminite, F = fusinite, Fb = fishbone, Zc = zooclasts, La = lamalginite, Sp = sporinite. See text for further information.*

## 388 5. Discussion

### 389 5.1 Spherulite formation

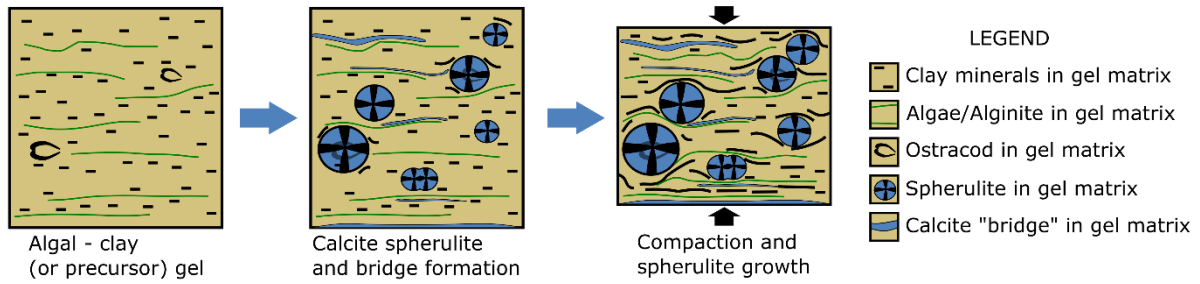
390 Spherulites form in a variety of depositional environments, from soil (e.g. Verrecchia  
391 et al., 1995) to travertine (e.g. Pentecost, 2005; Chafetz et al., 2018) to cave (e.g. Self and Hill,  
392 2003) to lake (e.g. Mercedes-Martín et al., 2017; Bischoff et al., 2020) and even reef settings  
393 (e.g. Sun et al., 2017). They are explained to form by microbial mediation (e.g. Verrecchia et  
394 al. 1995; Mercedes-Martín et al., 2016; Kirkham and Tucker, 2018; Wu et al., 2020) but also  
395 an abiotic origin has been proposed (e.g. Monaghan and Lytle, 1956; Oppenheimer, 1961;  
396 Verrecchia et al., 1995; Tosca and Wright, 2015). Depending on the formation processes and  
397 the conditions during formation, spherulite fabrics and/or organization will differ (e.g. Wright  
398 and Barnett, 2020 and Mercedes-Martín et al., 2021). Although their size and fabrics can reveal  
399 a lot about the formation processes, it is equally important to consider the matrix surrounding  
400 the spherulites. A common point for spherulite formation seems to be the growth in viscous  
401 media, where crystallization is slow (Tracy et al., 1998b). Although a viscous medium is  
402 strictly not a prerequisite, it strongly catalyzes spherulite growth (Beck and Andreassen, 2010;  
403 Shtukenberg et al., 2012). Bella and Garcia-Ruiz (1987), for example, reported on spherical  
404 growth in gel. Experiments point out that the growth of spherulites from solution happens when  
405  $Mg^{2+}$  and  $SO_4^{2-}$  are present. When either ion is absent, rhombohedra form (Tracy et al., 1998a).  
406 Thus chemistry and viscosity of the medium are considered to represent key parameters in  
407 spherulite formation, and leave their traces in the fabrics during lithification (e.g. Wright and  
408 Barnett, 2020).

409 In lacustrine settings several formational models are suggested (Gomes et al., 2020a),  
410 i.e. deposition in the water column or at the sediment water interface (Mercedes-Martin et al.,  
411 2017; Chafetz et al., 2018), or syndepositional to early diagenetic below the sediment-water  
412 interface (Dorobek et al., 2012; Saller et al., 2016; Wright and Barnett, 2019). In the latter case,

413 their formation requires the sediments to be still in an unconsolidated state (e.g. Herlinger et  
414 al., 2017; Gomes et al., 2020a).

415         The Aptian Barabalha spherulites display a large range of sizes even within a sample,  
416 so no sorting processes were involved. This would also conflict with the hydrodynamic energy  
417 needed and their presence in very fine-grained sediments. In addition, they occur in a clayey  
418 organic-rich matrix with the platy Mg-rich clay mineral and organic matter rich layers bended  
419 around the spherulites. Occasionally the spherulites even show compromised growth or lateral  
420 transition into crustiform calcites. Sulfate present as gypsum could be, given the at least  
421 temporary evaporitic conditions, primary or could be secondary, related to weathering of  
422 pyrite. Furthermore, samples rich in ostracods show spherulites that encapsulated some of the  
423 ostracods, thus the latter were present before spherulite growth. The whole assembly of these  
424 characteristics strongly point to a kind of chemically-driven concretionary/nodular radial  
425 growth around ostracods or micritic cores enabled by a viscous matrix or gel formed by organic  
426 matter from algae and clay or a precursor (Fig. 7). Pyrite and their weathered iron oxide  
427 equivalents are found inside the spherulites, but mainly concentrated at the spherulite-matrix  
428 interface, indicating a displacive growth. Although the clays do not show the typical autigenic  
429 cementation textures, based on their occurrence as films contouring ostracodes and spherulites,  
430 their formation is interpreted as in-situ, mimicking the depositional orientation, i.e. lamination,  
431 of their precursor (similar to Wright and Barnett, 2020). Due to the spherulite growth, plate  
432 minerals were pushed aside, but probably differential compaction between the matrix and  
433 spherulites during burial emphasized their anastomosing morphologies (Fig. 7).

434



435

436 *Figure 7: Calcite spherulite and brige formation in an algal – clay viscous matrix. Ostracods*  
 437 *can act as nucleation site. Due to the spherulite growth, plate minerals are pushed aside.*  
 438 *Differential compaction between the matrix and spherulites during burial emphasized their*  
 439 *anastomosing morphologies. Modified from Wright and Barnett (2017). Note also the*  
 440 *similarities with Herlinger et al. (2017). Of importance is that these authors consider*  
 441 *spherulite formation in the complete absence of organic matter.*

442

443 Similarities to the Pre-Salt spherulites are striking. The Barra Velha spherulites, unless  
 444 reworked, occur in two “*in-situ*” associations: loosely packed in Mg-silicate matrices and in  
 445 floatstone textures with the spherulites surrounded by porous matrices. The pores resemble  
 446 fenestrae, but are associated with dolomite bridges that mimic the shapes and orientations of  
 447 clay plates wrapped around grains (Wright and Barnett, 2017; 2020). While the latter authors  
 448 observed diagenetic dolomite bridges with a clay resemblance, in the Barbalha samples we  
 449 observe calcite bridges and the clays are still well-preserved (Figs. 2, 3 and 7). Wright (2012),  
 450 Tosca and Wright (2015) and Wright and Barnett (2020) consider a Mg-silicate gel as the  
 451 viscous medium in which the spherulites formed, that by lithification turns into Mg-silicates.  
 452 Dissolution of these Mg-silicates in turn could have been the cause of a significant part of  
 453 secondary porosity development in the Barra Velha carbonates (Tosca and Wright, 2015). The  
 454 clays in the Araripe spherulites are dominantly interstratified illite – montmorillonites and thus  
 455 differ from the tri-octahedral stevensites reported by Tosca and Wright (2015). Despite the  
 456 unique abundance and significance of stevensite, several other clay types have been found in

457 Pre-Salt subsurface reservoirs, e.g. kerolite-smectites and illite-smectites (Herlinger et al.,  
458 2017; Antonio, 2019). Tri-octahedral Mg-silicate stevensite (Abrahao and Warne, 1990) forms  
459 from Mg-rich lake waters with a pH >9 (Jones, 1986) and with possible magmatic  
460 hydrothermal activity (e.g. Tosca and Wright, 2015; Lima and De Ros, 2019). Montmorillonite  
461 and illite form in basins surrounded by rocks rich in Mg<sup>2+</sup>, Ca<sup>2+</sup> and Fe<sup>2+</sup> providing stagnant  
462 (reducing) alkaline water conditions under evaporative conditions, whereby illite forms under  
463 alkaline conditions and montmorillonite at even higher pH values (Reeves, 1968). For both  
464 cases, high pH conditions thus prevailed. This is in line with growth of carbonate spherulites  
465 in lacustrine conditions that has been proven to be restricted to specific pH conditions (pH 9.5-  
466 11; Garcia-Ruiz, 1998).

467         The high TOC of the spherulite matrix of the Barbalha spherulites differs from the  
468 observations for the Pre-Salt. The oil-prone alginate, however, would contribute to the  
469 hydrocarbons charging the reservoir (see also Schoellkopf and Patterson, 2000; Saller et al.,  
470 2016 and Spigolon et al., 2015). Given that hydrocarbon formation will be associated with  
471 carboxylic and other acids, decay of organic matter could explain part of the high secondary  
472 porosity in some of the Pre-Salt spherulites. For sure for the Barbalha spherulites, the organic  
473 matter contributed to the viscous medium in which the spherulites precipitated. Although that  
474 Tutolo and Tosca (2018) suggested that neither elevated concentration of organic acids, nor  
475 biological mediation, nor a gel matrix are required for calcite spherulites, Mercedes-Martin et  
476 al. (2016) show that the simplest mechanism to deposit significant spherical-radial calcite  
477 bodies is to begin with a strongly supersaturated solution that contains specific but  
478 environmentally common organic acids. Organic substances can lower the kinetic thresholds  
479 for nucleation, while maintaining a constant supersaturation during spherulite growth  
480 (Mercedes-Martín et al., 2021). In addition, based on experimental work from Wu et al. (2020),  
481 alginate turned out to be essential for the growth of spherulites. Bontognali et al. (2008),

482 considered their fine-grained cores to precipitate in Extracellular Polymeric Substances (EPS).  
483 Spherulites, in particular those with micritic nuclei, are suggested to have been nucleated by  
484 microbial mediation (e.g. Chafetz et al., 2018). Honeycomb textures were observed by SEM in  
485 the spherulite cores and plenty of alginate was observed in the rock matrices. Spherical shapes  
486 can be produced with or without the presence of bacterial cells within a gel medium (Chekroun  
487 et al., 2004; Turner and Jones, 2005), which in the Barbalha spherulites to large extent relates  
488 to EPS from algae, in line with experimental work from Wu et al. (2020) and Mercedes-Martín  
489 et al. (2021). Wright and Barnett (2017) pointed out that spherulitic carbonates are common  
490 features in microbial mats but are several orders of magnitude smaller than those seen in the  
491 Barra Velha (Sanchez-Navas et al., 2009; Spadafora et al., 2010; Wanas, 2012). A decreased  
492 spherulite size could be related to increased salinities (Mercedes-Martín et al., 2021). However,  
493 large-sized spherulites are observed in the Aptian Barabalha samples (Batateira bed) as  
494 discussed in this study. Even though their spherulite nucleation started from irregular grains or  
495 as tiny spheres in the organic-rich matrix, they resulted in very well-rounded, up to cm-scaled  
496 spherulites. In addition, Meister et al. (2011) show that calcite spheres, similar to those of the  
497 Araripe Basin and the Pre-Salt, only form from a solution with organic matter present, while  
498 spherical precipitation in abiotic experiments resulted in aragonite. Furthermore, not only the  
499 spheres but also the clays could have partially formed from the viscous medium rich in EPS  
500 (Fig. 7). Bischoff et al. (2020) also show the essential presence of EPS both in the formation  
501 of Mg-Si precipitates as well as for high-Mg calcite spherulites. In conclusion, the observations  
502 from the Batateira spherulites suggest that instead of considering microbial and clay or silica  
503 gel mediation as opposing theories, they should rather be considered as complimentary and/or  
504 cooperative in the genesis of spherulites. Considering the abiotic versus biotic formation, the  
505 presence and catalyzing effect of the alginate matrix is undeniable for the Barabalha  
506 spherulites.

507

## 508 5.2 *Water depth*

### 509 5.2.1 *Extra-spherulitite facies information*

510 Often lateral and vertical facies changes are used to interpret changes in the depositional  
511 system, including water depth. The abrupt transition from fluvial deposits to spherulite-  
512 bearing lacustrine lithologies is, however, structurally controlled. Scherer et al. (2015)  
513 considered epeirogenic regional movements as part of the sag basin during the post-rift stage  
514 responsible for the rapid increase in subsidence and creation of accommodation space. This  
515 resulted in a sudden relatively deep depression, with continued freshwater supply, most likely  
516 fed by the formerly existing river system. Renewed tectonic activity resulted in a very  
517 significant change in the drainage regime, base level, or both, resulting in an erosive  
518 unconformity and therefore a temporarily interrupted lake system (Assine et al., 2014;  
519 Fambrini et al., 2019). After a temporal riverine system, again a more persistent lacustrine  
520 system established that transitioned into the lacustrine carbonates of the Crato Formation, the  
521 main lacustrine phase in the Araripe Basin.

522 The transition from the upper lacustrine phase of the Barbalha formation towards the  
523 Crato laminites, consequently, is continuous and so the latter might hold crucial information  
524 regarding the water depth of the paleo-lake. In addition, the laminites possibly represent a  
525 better-exposed occurrence of the laminites observed as part of the Batateira beds. Thirdly, the  
526 oil shale facies with dark clay/organic-rich layers and nodular carbonates (Neumann et al.,  
527 2003; their Fig. 5) or nodular microfacies (Catto et al., 20016; their Fig. 6) or limestone with  
528 organic laminae (Cabral et al., 2019; their Fig. 6) might represent later occurring spherulite  
529 events as part of the Crato lacustrine system.

530 However, as summarized by Martill et al. (2007) the Crato paleo lake water has been  
531 considered as fresh (Maisey, 1990; 1996), hypersaline (Martill and Wilby, 1993) or brackish

532 (Bechly, 1998; Neumann et al., 2003) and the water depth as both, shallow (Maisey, 1990;  
533 Warren et al., 2017) and relatively deep (Martill and Loveridge, 2006). The depositional setting  
534 has been referred to as a lake or a lagoon (Martill, 1993), closed, or with a connection to the  
535 ocean in an epeiric sea setting (Arai, 2000; Assine et al., 2016, Varejao et al., 2016). Martill et  
536 al. (2007) and Barling et al. (2020) relate the remarkably low abundance of sedimentary  
537 patterns in the laminites to deposition below the storm wave base, i.e. to a water depth of 50 -  
538 200 m. Martill et al. (2008) was more careful in his conclusions stating it was “probably more  
539 than 30 m”. Varejão et al. (2020) highlight the importance on focusing on separate intervals of  
540 the Crato Formation, since they reflect changing water levels and hydrological connections.  
541 Considering that the Crato *Konservat-Lagerstätte* laminites are found across the Araripe Basin  
542 (~8000 km<sup>2</sup>) and based on a comparison of the maximum water depths of present-day saline  
543 analogue lakes (Table 3), suggests that the deepest water depth of the lake was at least 20 m.  
544 The large areal extension, however, also implies that most of it (i.e. average water depth; Table  
545 3) most likely must have been more shallow. Studies by Catto et al. (2016), Warren et al. (2017)  
546 and Varejao et al. (2019) interpret the association of stromatolites with halite hoppers, *Kinneyia*  
547 wrinkles, pseudo-columns, domes, MISS, thrombolites, preserved microbial mats, oncoids and  
548 breccias in the *Konservat-Lagerstätte* interval of the Crato Formation as shallow-water  
549 conditions (<10 m depth). While the Crato laminites are geographically continuous over the  
550 entire Araripe Basin (Fabin et al., 2018), with a minimum NW-SE extent of ~70 km  
551 (Heimhofer et al., 2010), the spherulite Batateira beds are known to be laterally continuous at  
552 distances most likely just above 20 km (Scherer et al., 2015). The reduced lake size and the  
553 fact that tectonic events were able to quickly change between the Barbalha lake and riverine  
554 setting, suggest a more shallow water depth compared to the Crato laminites, i.e. maximum  
555 water depth ~ 10 m. The average water depth consequently most likely was <10m.  
556



557 *Table 3: Non-exhaustive overview of (hyper)saline lake dimensions at arbitrary*  
 558 *moments. These data gathered from several sources (see reference column in addition to*  
 559 *Wikipedia) should be considered critically, especially given the strongly varying sizes of these*  
 560 *lakes, with many of them being remnants of larger lakes.*

	<b>Length (km)</b>	<b>Width (km)</b>	<b>Surface (km<sup>2</sup>)</b>	<b>Average depth (m)</b>	<b>Max depth (m)</b>	<b>References:</b>
<b>Owens lake (U.S.A.)</b>	28	16	451		0.91	Benson et al., 1996; Bacon et al., 2018
<b>Dead Sea (Israel - Jordan)</b>	50	15	605	199	298	Nof et al., 2012
<b>Salton sea (U.S.A.)</b>			889		13	Doede et al., 2020
<b>Lago Cuitzeo (Mexico)</b>	20	20	350	27	27	Mendoza et al., 2006
<b>Lake Corangamite (Australia)</b>			234	5	7	Williams, 1995; Tweed et al., 2009
<b>Lake Poopó (Bolivia)</b>			1000	1	2	Zola et al., 2006
<b>Great Salt Lake (U.S.A.)</b>	120	45	4400	4.9	14	Null and Wurtsbaugh, 2020
<b>Lake Urmia (Iran)</b>	140	55	5200		16	Schulz et al., 2020
<b>Aral Sea (Kazakhstan - Uzbekistan)</b>			17160	10	40	Burr et al., 2019
<b>Ebi lake (China)</b>			1070	1.4	2.8	Zhou et al., 2019
<b>Lake Assal (Afar depression)</b>	19	6.5	54	7.4	>40	Pérez and Chebude, 2017
<b>Lake Abert (U.S.A.)</b>	24	11	170	1.5	3.4	Senner et al., 2018
<b>Mono lake (U.S.A.)</b>	21	15	183	17	48	Hodelka et al., 2020
<b>Lake Walker (Canada)</b>	33	<10			280	Benson et al., 1991
<b>Lake Chad (Nigeria - Niger -</b>	50	30	1540	1.5	11	Pham-Duc et al., 2020

<b>Chad - Cameroon)</b>						
<b>Lake Bogoria (Kenya)</b>		3.5	36		12	Renaut et al., 2017; Kiage et al., 2020
<b>Lake Nakuru (Kenya)</b>	12	10	55	0.3	1.8	Kiage et al., 2020
<b>Lake Elementaita (Kenya)</b>			21		<1	Kiage et al., 2020

561 *5.2.2 Intra-spherulitite facies information*

562 The illite – montmorillonites of the spherulitite matrices are known to accumulate in  
563 low energy settings, not prone to exposure, desiccation or reworking. The relatively low quartz  
564 and feldspar contents corroborate the low contribution of detritals. In addition, the  
565 accumulation of organic matter is characteristic for low(er)-energy sub- to anoxic depositional  
566 areas within the lake. Lacustrine carbonates alternating with montmorillonite and illite clays,  
567 and containing evaporites are diagnostic for closed, chemically stratified, saline lakes in a semi-  
568 arid environment (Reeves, 1968). Because high-water level stages would have favored thermal  
569 water stratification and organic matter preservation, these factors can support a ‘deep’  
570 depositional scenario below the wave base. Alternatively, shallow restricted bays or lagoons  
571 could also provide the required low energetic conditions. Microbial mats in the photic zone  
572 independent of water depth can provide anoxic conditions due to internal stratification (as  
573 reported by Varejao et al., 2019 and Barling et al., 2020 for the Crato Formation).  
574 Perforations/borings into the spherulites (Fig. 3K) requires their formation to take place before  
575 substantial burying and in sufficiently shallow water conditions. In addition, the presence of  
576 gypsum and pyrite in the spherulite-bearing lithologies is an indication of shallow water  
577 conditions. The iron oxides (next to pyrite) at the spherulite-matrix contact are indeed  
578 interpreted as alteration products of pyrite. Concentric enrichment of pyrite has previously been  
579 reported from carbonate concretions in organic carbon-rich mudrocks (Caleman and Raiswell,  
580 1995) and are thought to reflect the activity of sulfate-reducing bacteria at shallow depth in

581 non-lithified sediments (Heimhofer et al., 2017). Anoxic conditions are ideal for pyrite  
582 formation via bacterial sulphate reduction when sulfates are present.

583 Relatively high C/N values at constant  $\delta^{13}\text{C}_{\text{org}}$  of the organic rich spherulites (Table 1),  
584 suggest denitrification (possibly associated to an Oceanic Anoxic Event – OAE; similar to  
585 Bastos et al., 2020), typical for low oxygen hypolimnetic waters or bottom sediments.  $\delta^{18}\text{O}$ -  
586  $\delta^{13}\text{C}$  signals are a function of the geographical and climatic setting, hydrology and the history  
587 of the waterbody (Talbot 1990). The relatively low  $\delta^{18}\text{O}$  of the studied spherulites should not  
588 be surprising given the freshwater input from the fluvial/deltaic systems related to the fast  
589 tectonic transition. The low  $\delta^{13}\text{C}$  of the spherulites could be related to the high organic matter  
590 content, at least if the latter got oxidized and dissolved. Not only the matrix containing the low  
591  $\delta^{13}\text{C}$  organic matter, but also the calcite spherulites show low  $\delta^{13}\text{C}$  signatures (Table 1). This  
592 can be related to water aging (Holmden et al., 1998), which is greatest in shallow, poorly  
593 circulated waters (Immenhauser et al., 2003), in line with the extra-facies arguments for a lake  
594 with a depth below 10m, at least at the time of spherulite formation.

595

### 596 5.3 Araripe Basin - South Atlantic paleo-connection?

597 Similarities with deposits from adjacent Basins have often been the cause for proposing  
598 a marine ingression into the Araripe Basin. Arai (2000), based on the presence of  
599 dinoflagelates, stated that a global eustatic sea-level rise in the late Early Cretaceous resulted  
600 in the episodic establishment of a highly restricted NW-SE trending epeiric seaway linking the  
601 intracratonic Araripe Basin either with the opening of the northern part of the South Atlantic  
602 Ocean (via the Reconcavo, Tucano and Jatoba Basin complex) or the Central Atlantic Ocean  
603 (via the Parnaíba or Potiguar basins). For the Ipubi and Romualdo Formations this connection  
604 to the sea has been confirmed based on paleontological and geochemical data (Dias-Brito,  
605 1992, 2000; Azevedo 2001, 2004; Rodrigues et al., 2020; Melo et al., 2020), but for the

606 Barbalha and Crato Formations, it has been an ongoing topic of debate. Arai (1999, 2011, 2012,  
607 2014) found marine forms of foraminifera and dinoflagellates in the Crato Formation, and  
608 together with radiolaria, molluscs, echinoids, crustaceans and fishes from other Brazilian  
609 continental Basins invokes a Tethyan connection. Assine et al. (2016) replying to Arai (2014)  
610 emphasized the necessity of a multidisciplinary approach and pointed out the critical role of  
611 the Araripe Basin in the evaluation of an interbasinal paleo-connection. Stratigraphic and  
612 paleocurrent data indicate three different large-scale drainage basins in the northeast of Brazil,  
613 implying a marine ingression from the SE for the Araripe Basin via the Tucano and Jatobá  
614 Basins (Assine et al., 2016). Although Arai (2016) rejected the drainage scenario, both agree  
615 on the connection between the Araripe, Tucano and Jatobá Basins. Varejão et al. (2016)  
616 confirmed the presence of the Crato Formation, or at least very similar deposits, in the Tucano  
617 Basin and at inselbergs in the Jatobá Basin. They show the same paleo-drainage flow, i.e. south  
618 to the South Atlantic Ocean (or giant lake at that time), also in line with the paleoflow of the  
619 Barbalha Formation (Assine et al., 2014, Scherer et al., 2015; Fambrini et al., 2019). Varejão  
620 et al. (2016, 2019) confirm the presence of the undulated laminites but also the so-called  
621 “nodular laminites” containing spherulites in the North Tucano Basin. Varejão et al. (2019)  
622 described calcite spherulites in the shallow water Amargosa Beds which are stratigraphically  
623 correlated to the Batateira Beds. In this way, paleontological, paleo-current and lithofacies data  
624 all seem to support either a connection along the different basins or the similarity in  
625 depositional environments across the basins and emphasize the strategic importance to look for  
626 onland equivalents of the Pre-Salt deep reservoir units.

627

#### 628 *5.4 Biotic influence before, during and after Barbalha spherulite formation*

629 Providing evidence of biotic influence on mineralization processes remains one of the  
630 biggest challenges in geology. Summarizing all the petrographic evidence of the Barbalha

631 spherulites, however, reveals the biotic influence before, during and after spherulite  
632 formation. As evidenced by the well-preserved lamalginate organic matter, microbial mats  
633 and/or their biopolymers accumulated. These mats and biopolymers kinetically accelerated  
634 calcite nucleation and sustained supersaturation for continued spherulite growth, as also  
635 confirmed from laboratory experiments (Mercedes-Martín et al., 2021). The different type of  
636 substrates and their composition suggest that nucleation was governed by chemically suitable  
637 substrates, e.g. calcite fragments, ostracods and other shells, and/or ionic concentration in  
638 organic matter, often preserved as micrite peloids (e.g. Riding et al., 2000). The concretionary  
639 growth by chemically-driven concentration, in essence abiotic, was mediated by the sustained  
640 supersaturation governed by the viscous mixed microbial mat - clay(-precursor) matrix.  
641 Microborings and the concentric enrichment of pyrite reveal the microbial activity after  
642 spherulite formation. Anoxic conditions, caused by the microbial mats, were ideal for pyrite  
643 formation via bacterial sulphate reduction. These spherulites consequently represent a,  
644 perhaps atypical, example of hybrid microbial-abiotic carbonates (Riding and Virgone, 2020).

645

#### 646 *5.5 The Araripe Aptian deposits as a Pre-Salt carbonate analogue*

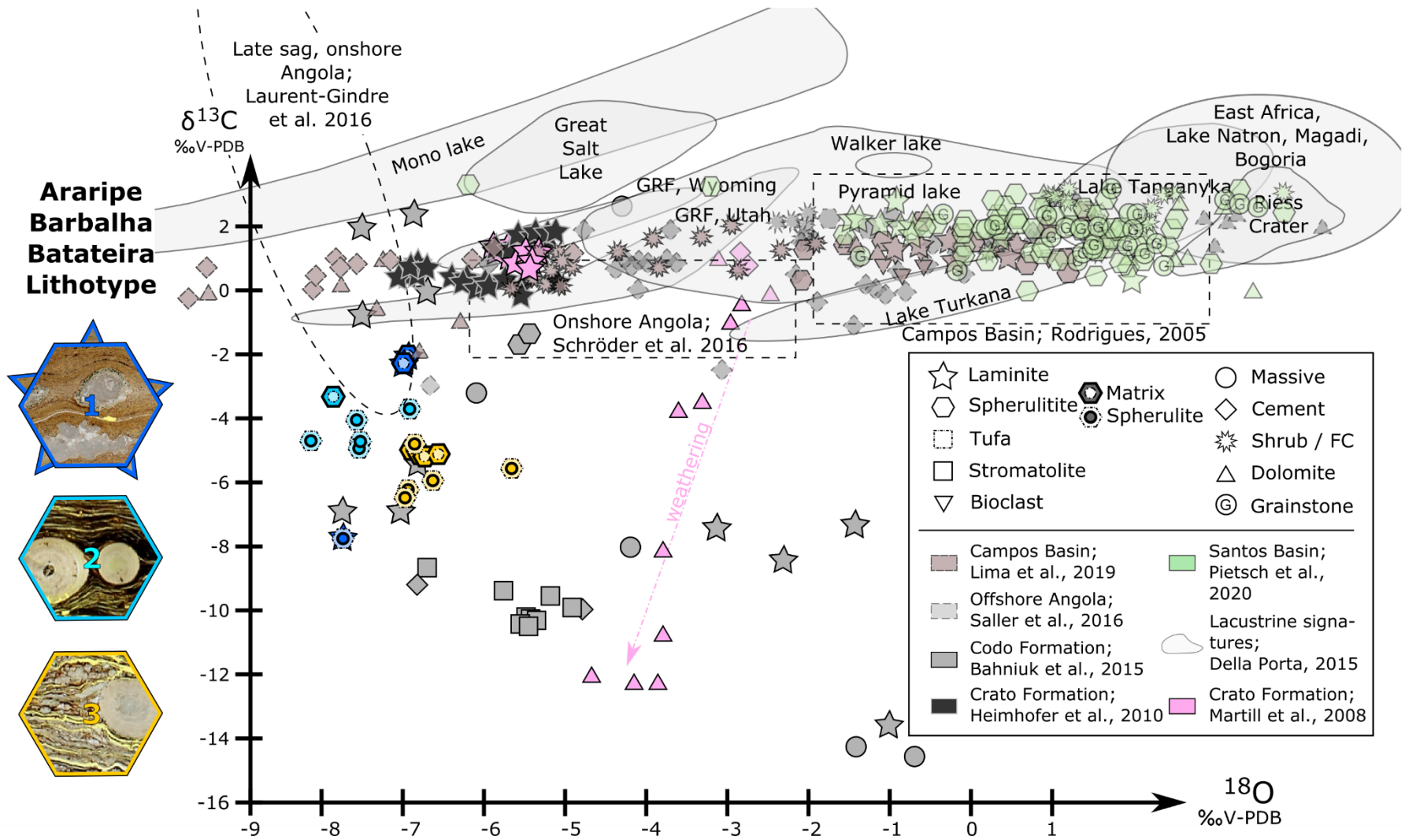
647 Lake deposystems are often controlled by highly complex hydrological factors  
648 reflecting the interaction of local climate, tectonics and the composition of the catchment  
649 geology (Wright and Rodriguez, 2018). Given that the Barbalha and geographically proximal  
650 Barra Velha and Macabu Formations are both lower Aptian deposits, they evidently possess  
651 several similarities with regard to their overall setting. The structural frame for all basins is the  
652 break-up of the Gondwana supercontinent associated with rifting and formation of horst-graben  
653 structures (Heine et al., 2013). Despite that the Pre-Salt deposits formed in the active rift basin  
654 ant that the Araripe basin developed as an intraplate intersection, they both are dominated by  
655 extension. In addition, the climatic conditions, i.e. semi-arid conditions (Doyle et al., 1982;

656 Lima and Perinotto, 1983; Ziegler et al., 2003), were the same. As such, not only the Araripe  
657 rock archive but also deposits in other inland basins, e.g. the Codo Formation (Bahniuk et al.,  
658 2005), can be considered as analogues. Dias (2005) described an epicontinental sea with a high  
659 continental influence and coastlines sectioned by valleys where in embayments, hypersaline,  
660 microbial communities formed microbialites. It is possible that temporarily or permanently  
661 connected volcanically influenced lakes in this rift setting are surrounded by the same rock  
662 types. The Barbalha (and Crato and Codo) Formation are part of the transitional sag phase,  
663 marked by thermal subsidence and huge shallow lake generation (e.g. Pietsch et al., 2018). The  
664 Cretaceous is known worldwide for its short-term eustatic sea-level changes (Ray et al., 2019).  
665 Large-scale marine incursions were occasional (Silva-Telles et al., 1994; Althaus et al., 2019;  
666 Rodrigues et al., 2020) and are recognized within these basins during the early Aptian (e.g.  
667 Crato Formation) but also for the overlying evaporites (Ipubi Formation in the Araripe Basin).  
668 The Barbalha laminites and spherulites are remarkably similar when comparing with Pre-Salt  
669 data and descriptions, not only to those reported on the Brazilian side of the South Atlantic but  
670 also in their West-African counterparts (Examples of spherulites in e.g. Herlinger et al., 2017;  
671 Farias et al., 2019; Lima and De Ros, 2019 and Wright and Barnett, 2020). Organic-rich  
672 mudstones, some with significant amounts of ostracods are reported, for example from the  
673 Kwanza deposits and are even considered the main source rocks in Pre-Salt (Saller et al., 2016).  
674 In addition, the type I organic matter of the Araripe spherulites (see also Spigolon et al., 2015)  
675 is in line with the dominance of type I in the Pre-Salt mudrocks (Mello et al., 1994; Mello,  
676 2006; Goldberg et al., 2017).

677 From a geochemical view, the Araripe and South-Atlantic Pre-Salt deposits show  
678 strongly different  $\delta^{18}\text{O}$  and  $\delta^{13}\text{C}$  signatures (Fig. 8). While the  $\delta^{13}\text{C}$  signatures of the Pre-Salt  
679 deposits are relatively constant, it should be noted that they show a high variation in  $\delta^{18}\text{O}$ . The  
680 nearly constant  $\delta^{13}\text{C}$  can be interpreted in terms of lake water  $\text{CO}_2$  in equilibrium with

681 atmospheric CO<sub>2</sub> (Pietsch et al., 2020). However, fractionation from magmatic CO<sub>2</sub>, or even  
682 CO<sub>2</sub> resulting from dissolution of marine carbonates at ~50 °C, would result in very similar  
683 signatures (Claes et al., 2015, Fig. 21), and should be considered when interpreting the stable  
684 isotope signatures of the PreSalt deposits. Both Pre-Salt laminites and spherulites show much  
685 higher δ<sup>18</sup>O signatures compared to their Araripe equivalents. These indicate a higher influence  
686 of meteoric waters both during deposition, as mentioned earlier, but also diagenesis for the  
687 Barbalha Formation. Indeed, the Araripe deposits have been exposed to meteoric alteration,  
688 recrystallisation and cementation, by which they would have inherited low δ<sup>18</sup>O signatures. On  
689 the other hand, diagenetic dolomite formation and dolomitisation are known to drive the δ<sup>18</sup>O  
690 to significantly higher values (e.g. McCormack et al., 2019) for varying δ<sup>13</sup>C. Dolomite  
691 formation is observed in the Araripe samples and some dykes were reported (Martill et al.,  
692 2008), with samples showing indeed a positive shift in δ<sup>18</sup>O (Fig. 7). In the Pre-Salt deposits,  
693 separating calcite from dolomite during sampling was not possible, however, Pietsch et al.,  
694 (2020) excluded the effect of mineralogy and facies on the stable isotopes. Lima et al. (2020),  
695 with supporting evidence from fluid inclusion microthermometry and radiogenic strontium  
696 isotopes, interpreted the δ<sup>18</sup>O-δ<sup>13</sup>C signatures as altered by hydrothermal fluids, with similar  
697 depositional and diagenetic processes over a regional scale, suggesting a vast lake or lakes that  
698 were hydro-geologically connected. It is exciting to imagine that the Araripe deposits were  
699 able to maintain the more pristine geochemical signatures as opposed to the strongly  
700 diagenetically altered Pre-Salt.

701  
702





704

705 *Fig. 8: Stable  $\delta^{18}\text{O}$ -  $\delta^{13}\text{C}$  isotopes of the lithofacies from the Barbalha Batateira*  
706 *deposits compared to literature data. The three lithotypes with spherulites of interest in this*  
707 *study are represented, including the (1) laminites with nodular spherulite and crust growth,*  
708 *(2) shaly organic-rich spherulitites containing spherulites with micritic cores, (3) shaly*  
709 *organic-rich spherulitites with a pronounced ostracod presence, also in the spherulites.*

710

## 711 **6. Conclusions**

712 The Aptian Araripe Barbalha spherulitites form an almost time-equivalent,  
713 geographically proximal, possibly sedimentological analogue for some of the Pre-Salt  
714 spherulites. Related to the horst-graben structural setting, the influence of surrounding  
715 lithologies, lateral and vertical facies continuity, the hypersaline lacustrine depositional  
716 environment, the biotic versus abiotic formation and petrophysical characteristics, the Barbalha  
717 Formation consequently might provide insight into the Pre-Salt reservoir rocks.

718 Based on the lithological characteristics of the Barbalha spherulites, information of  
719 surrounding lithologies and comparison with other lakes, the water depth of the paleolake at  
720 that time was less than 10 meter.

721 The presence of spherulites in organic, clay-rich lacustrine Aptian Araripe laminites  
722 suggests that microbial and clay mediation were complimentary processes in the genesis of the  
723 spherulites. Their nodular/concretionary growth encapsulated particles like e.g. ostracodes but  
724 acted displacively for sulfides and plate minerals.

725 The potential paleo-connection or at least similar depositional environments of the  
726 onland basins compared to the South Atlantic strengthen their use as analogues for  
727 depositional and early diagenetic conditions. However, important differences, for example with

728 regard to stable isotope geochemistry, should be further elucidated, taking into account the  
729 strongly contrasting diagenetic overprints.

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